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WS2: Avogadro's
Hypothesis Name: C
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Date: Hr: Goals: • I can
state Avogadro's
Hypothesis and the
evidence that led to this
hypothesis. In Unit 2,
you learned that the

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pressure of a gas is proportional to the Kelvin temperature (P T), when the volume and number of particles is held constant. Now consider equal volumes of two gases at the same ...

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- Counting Particles Too Small to See Using Avogadro's Hypothesis we are able to determine the number of molecules in macroscopic samples by weighing them. Unit 6 - Particles Having Internal Structure

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tools. The smallest unit

of a substance that

keeps all of the physical

and chemical properties

of that substance; it can

consist of one atom or

two or more atoms

bonded... 14

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this sample unit does
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substances Pure vs
Mixture (separation
techniques) Simple vs
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Unit 4 – Describing
Substance Objectives. 1.
Distinguish between a
pure substance and a
mixture by • properties.
• separation techniques.
• composition (macro-
and microscopically) 2.
Describe how to use
characteristic properties

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to separate the components of a mixture: • identify the separation technique

(name, equipment) •

identify the property used in the separation.

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require Java 1.5+ for
Windows, Linux, and
Mac OS X 10.4 and
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Bonding. Atoms form
bonds. In the first part

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of this unit, students learn about different types of bonds, principally ionic and covalent bonds. This unit focuses on recognizing why and how bonds form and the naming of the substances involved.

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Honors 1 U4 – review

v2. Name Date Pd.

Chemistry: Unit 4

–Concepts Review. To

prepare to do well on

the Unit 4 test, you

should assemble your

notes, the worksheets,

objectives sheet, and

review them, preferably

in a small group where

you can draw from each

other's understanding.

1.

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Quantum mechanics - Wikipedia

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